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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.034 wR factor = 0.093 Data-to-parameter ratio = 13.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

N'-(2,4-Dichlorobenzylidene)-2-methoxybenzohydrazide

In the crystal structure of the title compound, $C_{15}H_{12}Cl_2N_2O_2$, molecules are connected *via* weak intermolecular $C-H\cdots O$ hydrogen bonds, forming a two-dimensional network.

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Comment

Metal complexes based on Schiff bases have attracted much attention because they can be utilized as model compounds of active centers in various proteins and enzymes (Kahwa *et al.*, 1986; Santos *et al.*, 2001). Important to the understanding of the coordination potential of these ligands is a knowledge of ligand structure. Investigation of their crystal structures may provide useful information concerning their physical and chemical properties (Yu *et al.*, 2005; Deng *et al.*, 2005; Jing, Fan *et al.*, 2005*a*, 2005*b*; Jing, Wang *et al.*, 2005; Jing *et al.*, 2006; Guo *et al.*, 2006). In the present study, we report the synthesis and structure of the title compound, (I) (Fig. 1).



The Schiff base is approximately planar; the C8–C14 portion and the benzaldehyde group (C1–C7) are each planar, with r.m.s. deviations of 0.0148 (5) and 0.0017 (2) Å, respectively, and the dihedral angle between these planes is 4.23 (6)°. An intramolecular N–H···O hydrogen bond stabilizes the molecular conformation, while an intermolecular C–H···O hydrogen bond stabilizes the crystal structure (Table 2). The molecules associate in a two-dimensional pattern parallel to the (100) plane (Fig. 2).

Experimental

An anhydrous ethanol solution (50 ml) of 2,4-dichlorobenzaldehyde (1.75 g, 10 mmol) was added to an anhydrous ethanol solution of (50 ml) 2-methoxybenzhydrazide (1.66 g, 10 mmol) and the mixture was stirred at 330 K for 8 h under N₂, whereupon a colorless solution was obtained. The solvent was removed and the residue recrystallized from *N*,*N*-dimethylformamide. The product was isolated and then dried *in vacuo* to give pure (I) in 73% yield. Colorless single crystals suitable for X-ray analysis were obtained by slow evaporation of an *N*,*N*-dimethylformamide solution of (I).

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organic papers

Crystal data

 $\begin{array}{l} C_{15}H_{12}Cl_2N_2O_2\\ M_r = 323.17\\ \text{Triclinic, } P\overline{1}\\ a = 7.389 \ (3) \ \text{\AA}\\ b = 7.908 \ (3) \ \text{\AA}\\ c = 13.575 \ (5) \ \text{\AA}\\ \alpha = 80.877 \ (5)^{\circ}\\ \beta = 79.738 \ (5)^{\circ}\\ \gamma = 70.002 \ (5)^{\circ} \end{array}$

Data collection

Bruker SMART CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\rm min} = 0.901, T_{\rm max} = 0.939$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.034$ $wR(F^2) = 0.093$ S = 1.032545 reflections 191 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, $^{\circ}$).

O1-C8	1.208 (2)	C5-C6	1.396 (2)
N1-C7	1.274 (2)	C6-C7	1.458 (2)
N1-N2	1.3708 (19)	C8-C9	1.504 (2)
N2-C8	1.357 (2)	C9-C14	1.391 (2)
C1-C6	1.394 (2)	C9-C10	1.400 (2)
C7-N1-N2	115.05 (15)	O1-C8-N2	122.45 (16)
C8-N2-N1	121.24 (15)	O1-C8-C9	121.68 (16)
C1-C6-C7	122.09 (15)	N2-C8-C9	115.87 (15)
C5-C6-C7	121.22 (15)	C14-C9-C8	115.40 (16)
N1-C7-C6	121.32 (16)	C10-C9-C8	126.71 (15)

V = 729.3 (5) Å³

 $D_x = 1.472 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

 $\mu = 0.45 \text{ mm}^{-1}$

T = 293 (2) K

 $\begin{array}{l} R_{\rm int}=0.012\\ \theta_{\rm max}=25.0^\circ \end{array}$

Prism, colorless

 $0.22\,\times\,0.16\,\times\,0.14$ mm

4004 measured reflections

2545 independent reflections

 $w = 1/[\sigma^2(F_0^2) + (0.0509P)^2]$

where $P = (F_0^2 + 2F_c^2)/3$

+ 0.0881P]

 $\Delta \rho_{\rm max} = 0.17 \text{ e } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.25 \text{ e} \text{ Å}^{-3}$

 $(\Delta/\sigma)_{\rm max} < 0.001$

1968 reflections with $I > 2\sigma(I)$

Z = 2

Table 2

Hydrogen-bond geometry (Å, °).

$D - \mathbf{H} \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
N2-H2A···O2	0.86	1.96	2.627 (2)	134
$C15-H15C\cdots O1^{i}$	0.96	2.52	3.359 (3)	146

Symmetry code: (i) x, y + 1, z.

H atoms were included in calculated positions (C–H = 0.9–0.96 Å and N–H = 0.86 Å) and refined using a riding-model approximation, with $U_{iso}(H) = 1.2U_{eq}(C,N)$ or $1.5U_{eq}(methyl C)$.

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.



Figure 1

The molecular structure of (I), with displacement ellipsoids drawn at the 30% probability level.



Figure 2

Packing view of (I) along the a axis, showing inter- and intermolecular hydrogen bonds as dashed lines.

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